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This is the accepted version of a paper published in *Advanced Functional Materials*. This paper has been peer-reviewed but does not include the final publisher proof-corrections or journal pagination.

Citation for the original published paper (version of record):

Li, J., Naiini, M M., Vaziri, S., Lemme, M C., Östling, M. (2014)

Inkjet Printing of MoS₂.

Advanced Functional Materials, 24(41): 6524-6531

<http://dx.doi.org/10.1002/adfm.201400984>

Access to the published version may require subscription.

N.B. When citing this work, cite the original published paper.

Permanent link to this version:

<http://urn.kb.se/resolve?urn=urn:nbn:se:kth:diva-157597>

DOI: 10.1002/ ((please add manuscript number))

Full Paper

Inkjet Printing of MoS₂

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Keywords: inkjet printing, molybdenum disulfide (MoS₂), two-dimensional layered materials, photoluminescence, photodetectors

Abstract: A simple and efficient inkjet printing technology is developed for molybdenum disulfide (MoS₂), one of the most attractive two-dimensional layered materials. The technology effectively addresses critical issues associated with normal MoS₂ liquid dispersions (such as incompatible rheology, low concentration, and solvent toxicity) and hence can directly and reliably write uniform patterns of high-quality (5-7 nm thick) MoS₂ nanosheets at a resolution of tens of micrometers. The technology efficiency facilitates the integration of printed MoS₂ patterns with other components (electrodes), and hence allows fabricating various functional devices, including thin film transistors, photoluminescence patterns, and photodetectors, in a simple, massive and cost-effective manner while retains the unique properties of MoS₂. The technology has great potential in a variety of applications, such as photonics, optoelectronics, sensors and energy storage.

1. Introduction

Because of their unique properties, two-dimensional (2D) layered materials have attracted great interest in various research fields.^[1–4] Besides the most prominent graphene, molybdenum disulfide (MoS₂) is now widely regarded as a promising material for electronics and optoelectronics.^[5–9] Superior transistors^[5] (mobility >200 cm² V⁻¹ s⁻¹ and on/off current

ratio $\sim 10^8$) and very sensitive photodetectors^[6] have been demonstrated based on single-layer MoS₂ flakes. In addition, devices made from few-layer or multilayer MoS₂ flakes also exhibit excellent performance.^[7-9] In spite of the high device performance, these MoS₂ flakes are mainly obtained from the low-yield mechanical cleavage technique. Recently, chemical vapor depositions have been developed to produce large-scale MoS₂.^[10,11] However, as for the emerging macro-electronics^[12,13] which features cost efficiency, simple methods to fabricate solution-phase MoS₂ nanosheets are favorable in a variety of applications, such as sensors,^[14] photodetectors,^[15,16] and supercapacitors.^[17] Furthermore, patterning,^[18] especially direct patterning,^[17] of MoS₂ at micrometer dimensions has great significance in extending their applications. As a powerful direct writing technique,^[19] inkjet printing has received increasing interest in electronics.^[13,20] Great progress has recently been made in developing inkjet printing technology for graphene.^[21,22] In this work, we have established a reliable and efficient inkjet printing technology for MoS₂.

Similar to graphene, massive MoS₂ nanosheets can be conveniently obtained from sonicating bulk MoS₂ powders in an appropriate organic solvent, such as dimethylformamide (DMF) and N-methyl-pyrrolidone (NMP).^[1-3,23] With the pretreatments of bulk MoS₂ powders through grinding^[14] or intercalation by alkali ions,^[24,25] MoS₂ nanosheets can be effectively exfoliated in water. In general, however, none of the as-prepared MoS₂ dispersions are excellent for inkjet printing. They usually suffer from low viscosity (< 2 cP, as compared with the value about 10 cP favored by inkjet printing), short stable period, low concentration and/or solvent toxicity (or unsafety). The incompatible viscosity may reduce the controllability of droplet jetting, while unstable inks may cause flake aggregation during the printing course and block the nozzles. Both of them severely degrade the reliability of inkjet printing especially when small scale (down to the level of 100 μm) is required. The low concentration diminishes the printing efficiency, and the solvent toxicity causes environmental concerns. As a matter of fact, in spite of experimental attempts in the

field,^[14,23,24] no efficient inkjet printing technology has been demonstrated for mass production of MoS₂ patterns/devices in sub-mm scale. In this work we combine the solvent exchange^[21,26] and polymer stabilization^[27] techniques developed for graphene dispersions to address the above issues concerning inkjet printing of MoS₂. First, bulk MoS₂ powders are exfoliated in DMF through bath sonication and high-speed centrifugation to obtain thin and small MoS₂ nanosheets. The nanosheets are then stabilized by a small amount of polymers, ethyl cellulose (EC). Finally, the solvent DMF is exchanged with terpineol through distillation at a high volume ratio (DMF:terpineol = 20:1), by virtue of the significant difference in boiling point between DMF (boiling point ~ 150°C) and terpineol (boiling point ~ 220°C). Consequently, the solvent is exchanged from the low-viscosity (0.9 cP at 20 °C) and toxic DMF to high-viscosity (~40 cP at 20 °C) and nontoxic terpineol, and meanwhile the MoS₂ concentration is significantly increased. Mixing with ethanol, the rheology of MoS₂/terpineol dispersions may be tailored to be compatible with inkjet printing. As a result, all the critical challenges have been overcome, and efficient and reliable inkjet printing can be attained at the resolution of tens of micrometers (the limit of a normal inkjet printer).

2. Result and Discussion

Figure 1 shows the dilute MoS₂/DMF dispersion stabilized by EC (Figure 1a) and the final MoS₂/terpineol ink tailored with ethanol (Figure 1b). The color change of the dispersion from dark yellow (Figure 1a) to black (Figure 1b) suggests a significant increase in the MoS₂ concentration. Figure 1c indicates the absorbance spectra for the final MoS₂ ink diluted by 20 times with terpineol. The two characteristic peaks around 600 nm and 672 nm confirm the existence of MoS₂.^[2,14,25] Using the Lambert-Beer law and the extinction coefficient at 672 nm $\alpha_{672} = 3400 \text{ mL}/(\text{mg} \cdot \text{m})$,^[2] we estimate the MoS₂ concentration in our final ink to be ~ 0.1 mg/mL. The ink is stable (without occurrence of any visible sediments) for one week when staying still. During printing, the ink in the cartridge keeps on moving, which suppresses the

flake aggregation and extends the ink stability (without evidently clogging the nozzles) to 2~3 weeks. We also prepared MoS₂ inks with higher concentration through increasing the volume ratio of DMF: terpineol to 60:1, but the resulting ink was only stable for about 4 h (staying still). The attained concentration of 0.1 mg/mL and simultaneous stability period of one week have already approached the best quality of MoS₂ dispersions.^[2] We regard EC as an excellent stabilizer for MoS₂ inks because it can also improve the uniformity of printed patterns,^[21,22,27] prevent flake restacking,^[28] and importantly, be readily removed through a simple annealing^[21,22,27] so that the residual contamination is insignificant.

Using an inkjet printer (Dimatix DMP 2800), we have demonstrated excellent jetting performance for the MoS₂ inks. As shown in Figure S1, the MoS₂ ink droplets are well directed and constantly jetted out of all nozzles at an even velocity. The reliable jetting performance enables us to print various patterns at large scale. **Figure 2** shows the printed drop matrix, lines and film on untreated Si/SiO₂ substrates with a resolution (droplet diameter or line width) of ~ 80 μm. The as-printed patterns (Figure 2a-c) conform well to the designs, and after drying, they exhibit good uniformity (Figure 2d-f). Somewhat, however, the MoS₂ patterns suffer from the undesired coffee-ring effects (perimeter is thicker than interior, as clearly seen in Figure 2f). The coffee-ring effects, caused by the pinning of contact lines,^[29] are general phenomena during the drying of liquid drops. The contact line pinning is also observed for our MoS₂ inks (Movie V1 in Supporting Information). In contrast, with similar ink composition, the graphene inks can unpin the contact lines so that the coffee-ring effects are suppressed.^[21] This clear difference between graphene and MoS₂ inks is interesting and deserves further extensive studies.

The technology allows convenient and massive production of electronic devices based on MoS₂ nanosheets. **Figure 3a** shows the inkjet printed MoS₂ devices with printed silver electrodes on a SiO₂/Si substrate (the SiO₂ layer is 300 nm thick). The devices have been annealed at 450 °C in N₂ atmosphere to remove the stabilizing polymer. During annealing, the

N_2 atmosphere is needed to prevent MoS_2 from being oxidized. The Raman spectra of the printed MoS_2 films (Figure 3b) exhibit two characteristic peaks at 383.0 and 408.2 cm^{-1} , corresponding to the E_{2g}^1 and A_{1g} modes, respectively. The frequency distance between the two peaks is 25.2 cm^{-1} , suggesting that most MoS_2 flakes have more than 6 layers (a systematic study^[30] suggests the peak distance increases with the MoS_2 layer numbers and 6-layer MoS_2 flake has a peak distance $\sim 24.8 \text{ cm}^{-1}$). The morphology of the printed MoS_2 films are characterized through atomic force microscopy (AFM) and scanning electron microscopy (SEM). Both AFM (Figure 3c) and SEM (Figure 3d) images indicate that after annealing, the microstructure of the MoS_2 films appears in the form of percolation clusters, rather than uniformly distributed flakes. This is consistent with other observation of pinholes in films of centrifuged MoS_2 nanosheets.^[16] It is likely that phase separation^[31] (between MoS_2 and EC) takes place during annealing and hence form the percolation-cluster-like microstructure. The zoom-in view of the AFM images on the void region (Figure S2) shows many isolated individual MoS_2 flakes. From the cross-section profile analysis (Figure S2), we find most MoS_2 flakes to be about 5-7 nm thick with the lateral dimension ranging from 40 nm to 100 nm (there are a small amount of MoS_2 flakes larger than 100 nm, Figure S3). The flake thickness is consistent with the Raman spectra in Figure 3b (note 6 MoS_2 layers correspond to a thickness of $\sim 4.2 \text{ nm}$). The zoom-in view of the SEM images on the percolation clusters (Figure 3e) indicates randomly-distributed MoS_2 flakes (note during annealing, the N_2 atmosphere retards the decomposition of EC so that some polymer residue is still visible in Figure 3e). The annealed MoS_2 films actually constitute a double-level hierarchical percolation system:^[31,32] The first level is the apparent percolation clusters (Figures 3c,d) while each cluster comprises random MoS_2 flakes (Figure 3e) which form the second-level percolation. Double-level percolation systems usually possess significantly lower percolation thresholds (approximate to the products of the percolation thresholds for the two levels).^[31,32]

In our case, most MoS₂ flakes are locally concentrated to the percolation clusters so that the global spanning probability is evidently higher than that of a uniform film with the same amount of MoS₂. Consequently, much less MoS₂ flakes are required to percolate the systems, which is a desired merit for conductance improvement. To confirm the percolation effects, we study the dependence of the MoS₂ film conductivity σ on the number of printing passes N , as shown in Figure S4 where the drop spacing is 40 μm for the film printing. When $N \geq 15$, σ is almost independent of N which is expected for bulk-like materials. However, at small N region ($N < 15$), σ significantly increases with N as a power law $\sigma \sim (N - N_0)^n$ with $N_0 \approx 3.6$ and $n \approx 1.7$, suggesting the dominance of the percolation effects.^[23]

Photoluminescence (PL) is observed from our printed MoS₂ devices (**Figure 4**). Figure 4a indicates some printed MoS₂ devices under illumination of white light, while Figure 4b shows their PL image under an excitation laser with wavelength 532 nm. Usually only single-layer MoS₂ can generate strong PL thanks to the direct bandgap, and PL severely decreases with increasing MoS₂ layer numbers.^[10,25,33] Indeed, Figure 4b shows very weak PL from our printed MoS₂ (mostly > 6 layers) on the SiO₂/Si substrate. In contrast, however, the same MoS₂ on top of the silver electrodes generates very strong PL (the bright regions in Figure 4b). Note that this PL is also much stronger than that from the pure silver electrode. As shown in Figure 4c, the inkjet printed silver electrodes consist of many silver nanoparticles (mostly > 20 nm). Although these silver nanoparticles do not generate strong PL themselves, when attaching to the MoS₂ flakes (Figure 4d), they significantly enhance the PL from MoS₂.^[34] Figures 4e and 4f display the PL spectra at the MoS₂ region and the MoS₂/silver region, respectively. The two characteristic peaks around 630 nm and 680 nm confirm that the PL should be ascribed to MoS₂.^[10,25,33]

Furthermore, the silicon substrate is used as the back gate contact such that all devices are characterized as thin film transistors (TFTs), as shown in **Figure 5**. The transfer (Figure 5a) and output (Figure 5b) characteristics exhibit p-type gate modulation with a typical on/off

current ratio around 2. The large hysteresis in Figure 5a should imply the presence of a number of charge traps on the MoS₂ flakes.^[35] Although MoS₂ is intrinsically n-type semiconductor, liquid phase processing often introduces unintentional dopants or traps which may impact the electronic properties of MoS₂. As a result, low on/off current ratio (3 ~ 6) as well as p-type gate modulation are often observed in the literature for transistors based on liquid-exfoliated *individual* MoS₂ flakes.^[2,36,37] However, previously-demonstrated back-gate TFTs based on films of exfoliated MoS₂ nanosheets only exhibit negligible on/off current ratio (< 1.1).^[38] Our printed TFTs have comparable on/off current ratio to those based on individual liquid-exfoliated MoS₂ flakes, proving the efficiency of the printing technology. The current-voltage characteristics (Figure 5c) exhibit Ohmic behavior for all studied devices, while the resistance-length dependence (Figure 5d) indicates that the device resistance is almost proportional to the device length with a linear resistivity of $6.4 \times 10^8 \Omega/\mu\text{m}$. Hence the silver-MoS₂ contact resistance is negligible. The linear resistivity corresponds to a conductivity of $8.9 \times 10^{-5} \text{ S/m}$ (the device width is $\sim 350 \mu\text{m}$ and from the AFM image in Figure 3c we estimate the *average* film thickness to be no more than 50 nm). This value is even superior to that of the vacuum-filtered MoS₂ films ($3.6 \times 10^{-5} \text{ S/m}$).^[2]

The printed MoS₂ devices exhibit very rapid photoresponse. **Figure 6** shows the time-resolved photoresponse for a device with channel length $L = 13 \mu\text{m}$. The device was biased at gate voltage V_g and drain voltage V_d , and illuminated by a Halogen lamp (OSRAM HLX 64653) through a microscope objective lens, as schematically illustrated in Figure 6a. The power density of the illuminating light beam (white light) was 0.35 mW/mm^2 (Figure S5). During the measurements, the light was alternately switched on/off at an interval of 5 s. Following each on/off switch, the device current (at $V_g = 0 \text{ V}$ and $V_d = 1 \text{ V}$) swiftly jumps upwards/downwards and then returns toward the dark current level, resulting in a series of peaks and valleys. Surprisingly, the valleys may even fall below the dark currents (Figure 6b,c). The photoresponse significantly depends on the gate bias. For negative gate bias ($V_g = -$

5 V in Figure 6b), the photoresponse retains but the peaks do not appear for every switch on. For positive gate bias, the photoresponse attenuates; the signal becomes very weak at $V_g = V_d$ and vanishes when $V_g > V_d$. Such behavior has been confirmed with two other devices (Figure S6). Therefore, the optimal operating condition is around $V_g = 0$ V. Actually, excellent photoresponse can be achieved even without the gate bias, i.e., under floating gate (Figure S7). This feature simplifies the device fabrication and operation. However, one should note the optimal operating condition may vary with different substrates. For example, under floating gate, the printed MoS₂ devices on glass (Figure S8) exhibit very similar photoresponse to those on SiO₂/Si wafers under positive V_g . With the assistance of appropriate glass filters (THOR-labs, FGK01 Kit, Figure S9), the illumination can be tuned from white light to ultraviolet (UV), blue, red and infrared (IR) lights (the inset in Figure 6a shows the device illuminated by different visible lights). Under all tested illumination conditions ($V_g = 0$ V), the peak/valley retains in response to every switch on/off, even in the case of the lowest-power UV light. From the close-up view of the current-time curves in Figure 6c, the rise time for the peak and fall time for the valley are around 60 ms and 570 ms, respectively. The response time is comparable to that of the photodetectors based on monolayer MoS₂.^[6,39] In fact, the measured rise time (60 ms) is limited by the time resolution of our equipment and the true response time may be much shorter. Taking the average peak height (57 ± 11 pA for the white light at $V_g = 0$ V in Figure 6b) as the photocurrents, we estimate the photoresponsivity to be 36 ± 7 μ A/W (the device area is 4.6×10^{-3} mm² and the received power is about 1.6 μ W). Although this photoresponsivity is much lower than that of photodetectors based on individual MoS₂ flakes,^[6,9,39] it is comparable to those based on individual carbon nanotubes^[40] or graphene.^[41-43] In particular, it is also comparable to that of MoS₂ nanosheet films fabricated by the Langmuir-Blodgett (LB) method,^[16] implying that our inkjet printing processes do not induce obvious performance degradation.

The responsive peaks/valleys are highly reproducible (at $V_g = 0$ V or under floating gate) in all the printed photodetectors (Figures S6,S7). We have confirmed that these peaks/valleys are not caused by the light source (Figures S4,S10) or other electrical interferences (Figure S11), and should represent intrinsic response to the light switching. As shown in Figure 5c, a peak (valley) can be regarded as the combination of a fast rise (decay) and a sequent slow decay (rise) of the photocurrents. Similar behaviors are observed in other photodetectors based on MoS₂ nanosheets,^[16] although there a fast rise (decay) is followed by a slow rise (decay) of photocurrents. Similarly, we associate the current peaks and valleys with the presence of traps.^[16] As soon as the light is switched on, the fast current rise is ascribed to the initial equilibration of carrier generation and recombination.^[16] Afterwards, carriers are trapped and the current slowly decays. This forms the upward peaks. Upon switching off the light, the generation rate drops abruptly so that the predominant carrier recombination gives rise to a fast current fall. The current may even transiently fall below the initial dark current level. Gradually, the thermal release of the trapped carriers leads to a slow current increase. This forms the downward valleys. The traps may also serve as the generation-recombination centers which assist the generation/recombination process so that long-wavelength illumination (such as IR) can also be detected. At negative V_g (on state), there are more carriers and the trapping is faster. The photo-generated carriers may be trapped immediately, which prevents the appearance of some peaks upon illumination. At positive V_g (off state), the photo-generation rate may decrease with increasing V_g (likely because of the enhanced detrapping process) and hence the photoresponse attenuates or even vanishes. Therefore, all response behavior should result from the coupling between photo-generation/recombination and trapping/detrapping. These characteristics are in essence consistent with those of MoS₂ photodetectors fabricated by the LB method.^[16]

Finally, we would mention that although our MoS₂ inks are only stable for one week when staying still or 2~3 weeks when keeping moving during printing, the precipitated MoS₂ ink

can be recovered by a rough sonication for one hour. Comparable jetting (Figure S12) and electrical (Figure S13) performance to those of the fresh inks can be observed for the recovered MoS₂ ink which has even been stored for more than one year since it was initially prepared.

3. Conclusions

In conclusion, by employing the solvent exchange and polymer stabilization techniques, we have developed a reliable and efficient technology for inkjet printing of multilayer (> 6 layers, 5-7 nm thick) MoS₂ nanosheets. Conformal and uniform patterns can be printed with a resolution of about 80 μm (limited by the standard low-resolution printer and without any optimization for the substrate surface). Integrating the printed MoS₂ nanosheets with printed silver nanoparticles, we have demonstrated a simple method to significantly enhance the photoluminescence of multilayer MoS₂. The fully-printed MoS₂-based thin film transistors and photodetectors exhibit commendable performance verifying the retainability of MoS₂'s unique properties after printing. Consequently, the inkjet printing technology allows efficient, massive and cheap production of functional devices without inducing evident degradation of the MoS₂ materials. Further performance boost is anticipated once the general issues have been addressed concerning unintentional dopants/impurities in liquid-exfoliated MoS₂. We expect the technology will play an important role in a variety of applications, including photonics, optoelectronics, sensors and energy storage.

4. Experimental Section

MoS₂ exfoliation. First, a mixture of ~ 2 mg/ml MoS₂ powder (Sigma-Aldrich, product no. 69860) in DMF was sonicated (Branson 2510E-MTH bath ultrasonicator) for ~ 48 h. The resultant suspension was centrifuged at 10,000 rpm for 30 min to sediment thick flakes and the supernatant was harvested. The supernatant was centrifuged once more at 10,000 rpm for

30 min to ensure the removal of big flakes. Then, 0.25 mg/mL ethyl cellulose (viscosity 4 cP for 5%w/v in 80:20 toluene:ethanol, Sigma-Aldrich, product no. 200646) was added to the harvested supernatant and the dispersion was sonicated for ~1 h. Finally, another ~ 0.25 mg/mL ethyl cellulose (viscosity 22 cP for 5%w/v in 80:20 toluene:ethanol, Sigma-Aldrich, product no. 200697) was added to obtain a stable MoS₂/DMF dispersion.

Solvent exchange. 5 ml terpineol (Sigma-Aldrich, product no. 86480) was added to 100 ml MoS₂/DMF dispersion. DMF was exchanged by terpineol through a vacuum distillation process using a Rotary Evaporator (Rotavapor R-205, BÜCHI Labortechnik AG). DMF began to evaporate when the dispersion was heated to 80 °C and the pressure was reduced to ~30 mbar. After DMF was boiled off, the remaining MoS₂/terpineol dispersion was harvested. Compared with the original DMF dispersion, MoS₂ in the final terpineol dispersion was concentrated by ~20 times. Prior to printing, the MoS₂/terpineol dispersion was mixed with ethanol at the volume ratio of 3:1 to tailor the viscosity and surface tension for inkjet printing.

Concentration estimation. In order to estimate the final MoS₂ concentration, 50 µL of the final ink was diluted with terpineol to 1 mL and the optical absorbance A was measured by a Varian Cary 100 Bio UV-Vis Spectrophotometer, as presented in Figure 1c. Note around the characteristic peak position ($\lambda = 672$ nm), $A \approx 0.18$. Through the Lambert-Beer law $A/l = \alpha_{672}C$ ($l = 1$ cm is the cell length) and the absorption coefficient $\alpha_{672} = 3400$ mL/(mg·m),^[2] the MoS₂ concentration C in the diluted ink is estimated to be ~5.3 µg/mL. Therefore, the MoS₂ concentration in the final ink (20 times higher) should be ~0.1 mg/mL. Note as declared in Supporting Information of Ref. [2], the absorption coefficient α_{672} may vary with solvents. Then the value of 0.1 mg/mL should only be regarded as a rough estimation of the MoS₂ concentration in the final inks.

Device Fabrication. The patterns and devices were fabricated using a commercial piezoelectric Dimatix Material Printer (DMP 2800, Dimatix-Fujifilm Inc.) equipped with a 10

pL cartridge (DMC-11610). Our formulated MoS₂ inks and the commercial silver inks (Cabot Conductive Ink CCI-300, Cabot Corporation) were used. The lines and film in Figure 2 were printed with the drop spacing of 40 μm and 30 μm , respectively. The devices in Figure 3 were printed on silicon wafers capped with a thermally-grown SiO₂ layer (300 nm thick). Before printing, the wafers were cleaned with isopropanol. First, silver inks were printed (drop spacing of 20 μm , one printing layer, substrate at room temperature) as the electrodes and baked at 150 °C for 30 min. Then, MoS₂ inks were printed (drop spacing of 20 μm , 3 printing layers, substrate at 40 °C) as the channels. Finally, the devices were annealed at N₂ (100 sccm) atmosphere: The temperature ramped at 10 °C /min to 160 °C, retained at 160 °C for 20 min, then ramped at 10 °C /min to 450 °C and eventually retained at 450 °C for 1 h. Before electrical characterization, the annealed devices were covered by a 50-nm-thick Al₂O₃ layer through atomic layer deposition at 200 °C for passivation.

Electrical characterization. All electrical properties were characterized using Keithley SCS4200 parameter analyzer with a probe station. For photoresponse measurement, the devices are illuminated by a Halogen lamp (OSRAM HLX 64653) through the microscope objective lens. The UV, blue, red, and infrared lights were produced through filtering the white light with the THOR-labs colored glass filters (see Figure S9 for details). The switch of the illuminations was operated manually.

Supporting Information

Supporting Information is available from the Wiley Online Library or from the author.

Acknowledgements

The authors thank Ingemar Petermann (Swedish ICT Acreo), Ilya Sytjugov, Fatemeh Sangghaleh, and Miao Zhang for kind helps in device measurement and characterization. Support and sponsorship by the European Research Council through the Advanced Investigator Grant OSIRIS (No. 228229) and the Proof of Concept Grant iPUBLIC (No. 641416), the Swedish Research Council through the grant iGRAPHENE, and the Göran Gustafsson Foundation through the Young Researcher Prize (No. 1415 B) are gratefully acknowledged. M. C. Lemme acknowledges support from the European Research Council

through the Starting Grant InteGraDe (No. 307311) and the German Research Foundation (LE 2440/1-1).

Received: ((will be filled in by the editorial staff))

Revised: ((will be filled in by the editorial staff))

Published online: ((will be filled in by the editorial staff))

- [1] Q. H. Wang, K. Kalantar-Zadeh, A. Kis, J. N. Coleman, M. S. Strano, *Nat. Nanotechnol.* **2012**, *7*, 699–712.
- [2] J. N. Coleman, M. Lotya, A. O'Neill, S. D. Bergin, P. J. King, U. Khan, K. Young, A. Gaucher, S. De, R. J. Smith, I. V. Shvets, S. K. Arora, G. Stanton, H.-Y. Kim, K. Lee, G. T. Kim, G. S. Duesberg, T. Hallam, J. J. Boland, J. J. Wang, J. F. Donegan, J. C. Grunlan, G. Moriarty, A. Shmeliov, R. J. Nicholls, J. M. Perkins, E. M. Grieveson, K. Theuvsen, D. W. McComb, P. D. Nellist, V. Nicolosi, *Science* **2011**, *331*, 568–571.
- [3] V. Nicolosi, M. Chhowalla, M. G. Kanatzidis, M. S. Strano, J. N. Coleman, *Science*. **2013**, *340*, 1226419.
- [4] J. Li, M. Östling, *Phys. Rev. E* **2013**, *88*, 012101.
- [5] B. Radisavljevic, A. Radenovic, J. Brivio, V. Giacometti, A. Kis *Nat. Nanotechnol.* **2011**, *6*, 147–150.
- [6] O. Lopez-Sanchez; D. Lembke, M. Kayci, A. Radenovic, A. Kis, *Nat. Nanotechnol.* **2013**, *8*, 497–501.
- [7] S. Kim, A. Konar, W.-S. Hwang, J. H. Lee, J. Lee, J. Yang, C. Jung, H. Kim, J.-B. Yoo, J.-Y. Choi, Y. W. Jin, S. Y. Lee, D. Jena, W. Choi, K. Kim, *Nat. Commun.* **2012**, *3*, 1011.
- [8] H. S. Lee, S. Min, Y. Chang, M. K. Park, T. Nam, H. Kim, J. H. Kim, S. Ryu, S. Im, *Nano Lett.* **2012**, *12*, 3695–3700.
- [9] D.-S. Tsai, K.-K. Liu, D.-H. Lien, M.-L. C.-F. Tsai, Kang, C.-A. Lin, L.-J. Li, J.-H. He, *ACS Nano* **2013**, *7*, 3905–3911.
- [10] Y.-H. Lee, X.-Q. Zhang, W. Zhang, M.-T. Chang, C.-T. Lin, K.-D. Chang, Y.-C. Yu, J. T.-W. Wang, C.-S. Chang, L.-J. Li, T.-W. Lin, *Adv. Mater.* **2012**, *24*, 2320–2325.
- [11] Y. Zhan, Z. Liu, S. Najmaei, P. M. Ajayan, J. Lou, *Small* **2012**, *8*, 966–971.
- [12] D. Voss, *Nature* **2000**, *407*, 442–444.
- [13] R. F. Service, *Science* **1997**, *278*, 383–384.
- [14] Y. Yao, L. Tolentino, Z. Yang, X. Song, W. Zhang, Y. Chen, C. Wong, *Adv. Funct. Mater.* **2013**, *23*, 3577–3583.

- [15] K.-G. Zhou, N.-N. Mao, H.-X. Wang, Y. Peng, H.-L. Zhang, *Angew. Chem. Int. Ed. Engl.* **2011**, *50*, 10839–10842.
- [16] G. Cunningham, U. Khan, C. Backes, D. Hanlon, D. McCloskey, J. F. Donegan, J. N. Coleman, *J. Mater. Chem. C* **2013**, *1*, 6899–6904.
- [17] L. Cao, S. Yang, W. Gao, Z. Liu, Y. Gong, L. Ma, G. Shi, S. Lei, Y. Zhang, S. Zhang, R. Vajtai, P. M. Ajayan, *Small* **2013**, *9*, 2905–2910.
- [18] C. L. Stender, E. C. Greyson, Y. Babayan, T. W. Odom, *Adv. Mater.* **2005**, *17*, 2837–2841.
- [19] D. B. Chrisey, *Science* **2000**, *289*, 879–881.
- [20] J. Li, T. Unander, A. López Cabezas, B. Shao, Z. Liu, Y. Feng, E. Bernales Forsberg, Z.-B. Zhang, I. Jögi, X. Gao, M. Boman, L.-R. Zheng, M. Östling, H.-E. Nilsson, S.-L. Zhang, *J. Appl. Phys.* **2011**, *109*, 084915.
- [21] J. Li, F. Ye, S. Vaziri, M. Muhammed, M. C. Lemme, M. Östling, *Adv. Mater.* **2013**, *25*, 3985–3992.
- [22] E. B. Secor, P. L. Prabhuram, K. Puntambekar, M. L. Geier, M. C. Hersam, *J. Phys. Chem. Lett.* **2013**, *4*, 1347–1351.
- [23] D. J. Finn, M. Lotya, G. Cunningham, R. J. Smith, D. McCloskey, J. F. Donegan, J. N. Coleman, *J. Mater. Chem. C* **2014**, *2*, 925–932.
- [24] J. Zheng, H. Zhang, S. Dong, Y. Liu, C. T. Nai, H. S. Shin, H. Y. Jeong, B. Liu, K. P. Loh, *Nature Commun.* **2013**, *5*, 2995.
- [25] G. Eda, H. Yamaguchi, D. Voiry, T. Fujita, M. Chen, M. Chhowalla, *Nano Lett.* **2011**, *11*, 5111–5116.
- [26] J. Li, F. Ye, S. Vaziri, M. Muhammed, M. C. Lemme, M. Östling, *Carbon* **2012**, *50*, 3113–3116.
- [27] Y. T. Liang, M. C. Hersam, *J. Am. Chem. Soc.* **2010**, *132*, 17661–17663.
- [28] J. Li, M. Östling, *Crystals* **2013**, *3*, 163–190.
- [29] R. D. Deegan, O. Bakajin, T. F. Dupont, *Nature* **1997**, *389*, 827–829.
- [30] C. Lee, H. Yan, L. E. Brus, T. F. Heinz, J. Hone, S. Ryu, *ACS Nano* **2010**, *4*, 2695–2700.
- [31] K. Levon, A. Margolina, A. Z. Patashinsky, *Macromolecules* **1993**, *26*, 4061–4063.
- [32] J. Li, B. Ray, M. A. Alam, M. Östling, *Phys. Rev. E* **2012**, *85*, 021109.
- [33] A. Splendiani, L. Sun, Y. Zhang, T. Li, J. Kim, C.-Y. Chim, G. Galli, F. Wang, *Nano Lett.* **2010**, *10*, 1271–1275.

- [34] K. Zhou, Y. Zhu, X. Yang, J. Zhou, C. Li, *ChemPhysChem* **2012**, *13*, 699–702.
- [35] W. Kim, A. Javey, O. Vermesh, Q. Wang, Y. Li, H. Dai, *Nano Lett.* **2003**, *3*, 193–198.
- [36] K. Lee, H.-Y. Kim, M. Lotya, J. N. Coleman, G.-T. Kim, G. S. Duesberg, *Adv. Mater.* **2011**, *23*, 4178–4182.
- [37] Z. Zeng, Z. Yin, X. Huang, H. Li, Q. He, G. Lu, F. Boey, H. Zhang, *Angew. Chem. Int. Ed.* **2011**, *50*, 11093–11097.
- [38] Q. He, Z. Zeng, Z. Yin, H. Li, S. Wu, X. Huang, H. Zhang, *Small* **2012**, *8*, 2994–2999.
- [39] Z. Yin, H. Li, H. Li, L. Jiang, Y. Shi, Y. Sun, G. Lu, Q. Zhang, X. Chen, H. Zhang, *ACS Nano* **2012**, *6*, 74–80.
- [40] M. Freitag, J. C. Tsang, A. Bol, D. Yuan, J. Liu, P. Avouris, *Nano Lett.* **2007**, *7*, 2037–2042.
- [41] F. Xia, T. Mueller, Y.-M. Lin, A. Valdes-Garcia, P. Avouris, *Nat. Nanotechnol.* **2009**, *4*, 839–843.
- [42] M. C. Lemme, F. H. L. Koppens, A. L. Falk, M. S. Rudner, H. Park, L. S. Levitov, C. M. Marcus, *Nano Lett.* **2011**, *11*, 4134–4137.
- [43] F. Xia, T. Mueller, R. Golizadeh-Mojarad, M. Freitag, Y. Lin, J. Tsang, V. Perebeinos, P. Avouris, *Nano Lett.* **2009**, *9*, 1039–1044.

Figure 1. MoS₂ inks. (a) Photograph of the MoS₂/DMF dispersion before distillation. The MoS₂ nanosheets have been stabilized by EC. (b) Photograph of the final MoS₂/terpineol ink. As compared with the initial MoS₂/DMF dispersion in (a), the MoS₂/terpineol ink is first concentrated by 20 times via distillation and then tailored by ethanol at the volume ratio of terpineol:ethanol = 3:1. (c) Absorbance spectra for the MoS₂ ink in (b) diluted by 20 times with terpineol.

Figure 2. Optical micrographs of inkjet printed MoS₂ patterns. (a) As-printed droplet matrix. (b) As-printed lines. (c) As-printed film. (d), (e) and (f) are dried patterns of (a), (b) and (c), respectively. All scale bars are 100 μm .

Figure 3. Inkjet printed MoS₂ devices on SiO₂/Si substrates (the SiO₂ layer is 300 nm thick). (a) Optical micrograph of the printed devices. The vertical strips are inkjet printed silver electrodes, and the horizontal line (~ 350 μm wide) is the printed MoS₂ channel. For electrical passivation, the whole devices have been covered by a layer of 50-nm-thick Al₂O₃ via atomic layer deposition which makes the substrates appear yellow. All morphology studies were performed without the Al₂O₃ deposition. (b) Raman spectra, (c) AFM image, (d) SEM image and (e) zoom-in SEM image on the MoS₂ (channel) region in (a).

Figure 4. Photoluminescence of printed MoS₂ on SiO₂/Si substrates. (a) Optical micrograph of the printed devices under illumination of white light. (b) PL image of the devices in (a) under illumination of an excitation laser (wavelength 532 nm). (c,d) SEM images of (c) silver electrodes and (d) MoS₂ on top of silver electrodes. (e,f) PL spectra of (e) MoS₂ on SiO₂/Si substrates and (f) MoS₂ on top of silver electrodes. The PL image (b) and spectra (e,f) were measured after an optical filter (longpass > 560 nm). Scale bars are 100 μm in (a) and (b), and 100 nm in (c) and (d).

Figure 5. Electrical characterization of the printed MoS₂ TFTs. (a) Transfer characteristics for a printed MoS₂ TFT ($L = 13 \mu\text{m}$). For each measurement, the gate voltage sweep is from -30 V to 30 V to -30 V. For the second sweep, the cable to the gate contact is physically disconnected. The invariant drain currents confirm the gate modulation in other sweeps. (b) Output characteristics (drain current-drain voltage curves) at different gate bias for the device in (a). (c) Drain current-drain voltage curves at $V_g = 0 \text{ V}$ for devices with different channel lengths. For each measurement, the drain voltage sweep is from -10 V to 10 V to -10 V. (d) Dependence of device resistance ($V_g = 0 \text{ V}$) on the channel length. The symbols are experimental data and the line is the linear fitting.

Figure 6. Photoresponse of a printed MoS₂ device ($L = 13 \mu\text{m}$). (a) Illustration of the MoS₂ device structure under illumination. Upper-right inset: Optical micrographs of the device under different illuminations of white (left), blue (middle) and red (right) lights. The scale bar is $100 \mu\text{m}$. (b) Time-resolved photoresponse of the device under different gate bias V_g and/or different illuminations. The illuminations are produced by tuning the white light with appropriate THOR-labs colored glass filters (see Figure S5 for details). The lights are on (off) in the shaded (unshaded) intervals. Each interval is 5 s. (c) A close-up view of the photoresponse to white light at $V_g = 0$ V [the topmost one in (b)]. For all measurements, $V_d = 1$ V.

An efficient inkjet printing technology is developed for direct and reliable writing of uniform patterns of high-quality MoS₂ nanosheets at a resolution of tens of micrometers. It provides a facile method for massive and cost-effective production of a variety of electronic and photonic devices, such as thin film transistors, photoluminescence patterns, and photodetectors.

Keywords: inkjet printing, molybdenum disulfide (MoS₂), two-dimensional layered materials, photoluminescence, photodetectors

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Inkjet Printing of MoS₂